


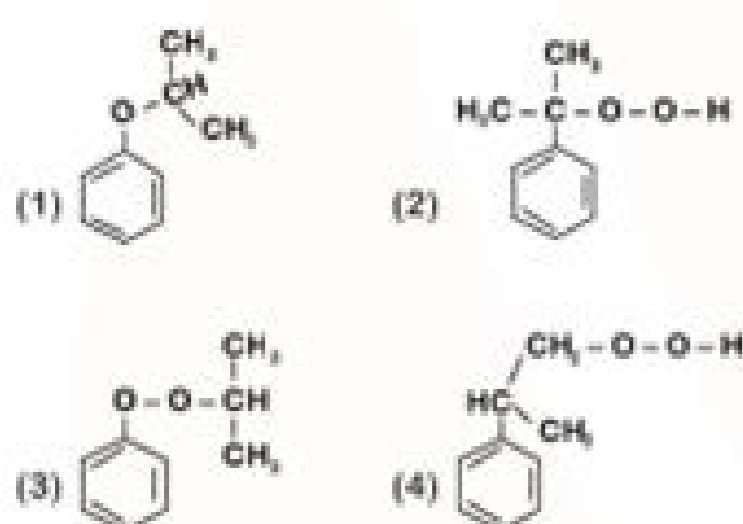
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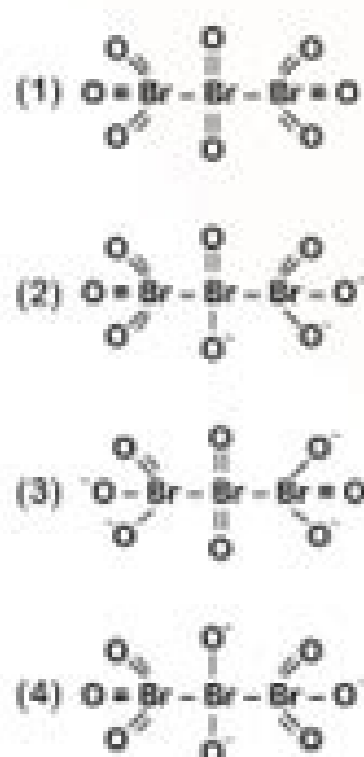


46. The number of sigma (σ) and pi (π) bonds in pent-2-en-4-yne is
- 10 σ bonds and 3 π bonds
 - 8 σ bonds and 5 π bonds
 - 11 σ bonds and 2 π bonds
 - 13 σ bonds and no π bonds

47. The structure of intermediate A in the following reaction, is

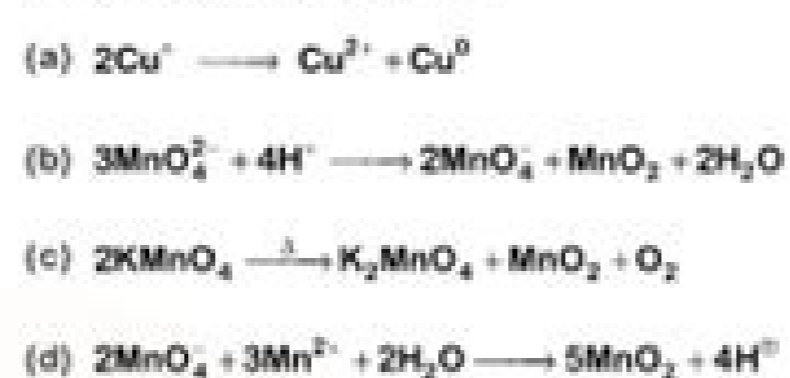


48. The correct structure of tribromooctaoxide is



49. 4d, 5p, 5f and 6p orbitals are arranged in the order of decreasing energy. The correct option is
- 5f > 6p > 5p > 4d
 - 6p > 5f > 5p > 4d
 - 6p > 5f > 4d > 5p
 - 5f > 6p > 4d > 5p

50. Which of the following reactions are disproportionation reaction?



Select the correct option from the following

- (a) and (b) only
- (a), (b) and (c)
- (a), (c) and (d)
- (a) and (d) only

51. Under isothermal condition, a gas at 300 K expands from 0.1 L to 0.25 L against a constant external pressure of 2 bar. The work done by the gas is

(Given that 1 L bar = 100 J)

- 30 J
- 5 kJ
- 25 J
- 30 J

52. Among the following, the one that is not a green house gas is

- Nitrous oxide
- Methane
- Ozone
- Sulphur dioxide

53. For the cell reaction



$E_{\text{cell}}^{\circ} = 0.24 \text{ V}$ at 298 K. The standard Gibbs energy ($\Delta_r G^{\circ}$) of the cell reaction is :

- 46.32 kJ mol⁻¹
- 23.16 kJ mol⁻¹
- 46.32 kJ mol⁻¹
- 23.16 kJ mol⁻¹

54. Enzymes that utilize ATP in phosphate transfer require an alkaline earth metal (M) as the cofactor. M is :

- Be
- Mg
- Ca
- Sr



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